STEREOSPECIFIC CYCLOPROPANE RING-OPENING OF PETROSTEROL.

A POSSIBLE BIOMIMETIC PROCESS.

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<u>Abstract</u>: Acid-catalyzed isomerization of the petrosterol side chain (<u>1</u>) proceeds stereospecifically to yield the naturally occurring 26-dehydro-25-epiaplysterol side chain (<u>2</u>); in addition, a 1.5-hydride shift leading to 22-dehydro-25-epiaplysteryl acetate (3) has also been observed.

Our postulate¹ that biosynthetic C-alkylation may also occur by isomerization of a cyclopropane intermediate together with our recent progress² on the acid-catalyzed ring-opening of steroidal cyclopropanes prompted us to examine both the stereo- and the regio-specificity of this process with a naturally occurring cyclopropane. For this purpose, we selected the acetate <u>1</u> of the sponge sterol petrosterol³ (isolated from Petrosia ficiformis^{3a}), since one of the possible products of acid-catalyzed² ring-opening should be the recently isolated⁴ "26-dehydroaplysterol" (cf. <u>2</u>) of unknown configuration at C-25. Elucidation of its C-25 stereochemistry is important in order to demonstrate a conceivable biosynthetic relation between <u>1</u> and <u>2</u>. We now report a successful resolution of both problems, which increases the likelihood that the postulated^{1,5} biosynthetic intermediacy of cyclopropanes operates in nature.

Petrosteryl acetate (<u>1</u>; m.p. 109-110^oC (MeOH); m/z 394.3596, 100, M-HOAc)⁶ was treated in the usual manner² with gaseous HCl in acetic acid for 150 min at room temperature. The complex reaction mixture was separated by reverse-phase HPLC (two Altex-Ultrasphere columns in series with absolute MeOH (3 mL/min) as the mobile phase) and 11 fractions, yielding together over 80%, were collected and each analyzed by mass spectrometry. All short retention time fractions (<135 min) show acetoxylated or chlorinated side chain fragments in their mass spectra which are due to molecules generated by nucleophilic ring-opening of the initially protonated steroidal cyclopropane.² The two "true" isomers <u>2</u> and <u>3</u>, having both retention times >150 min on HPLC, were isolated in 23% yield. The ¹H-NMR spectrum (see Table) of the main component (m/z 394.3587, 100, M-HOAc)⁶ showed it to be the anticipated isomerization product <u>2</u> which was identical with "26dehydroaplysterol" isolated⁴ from the sponge Petrosia ficiformis. The constitution of the side chain of the minor component <u>3</u> (m/z 394.3599, 100, M-HOAc)⁶ was established initially by comparison of its ¹H-NMR spectrum (see Table) with that of crinosteryl acetate (<u>4</u>)² which differs

compd	C-18	C-19	C-21	C-27	C-28	C-29
<u>1</u> (24R,25R,26R) ^b	0.680	1.019	0.920 (6.51)	-	0.889 (6.70)	1.007 (6.21)
<u>2</u> (24R,25R) ^C	0.674	1.017	0.902 (6.44)	0.930 (6.82)	0.793 (6.64)	-
<u>3</u> (22E,24S,25R) ^d	0.690	1.020	1.002 (7.0)	0.794 (6.75)	0.882 (7.0)	0.848 (7.35)
<u>4</u> (22E,24S) ^e	0.690	1.019	1.002 (6.68)	0.817 (6.63)	0.909 (6.78)	-
<u>5</u> (24R,25R) ^f	0.662	0.921	1.031 (6.48)	0.836 (6.57)	0.848 (6.69)	0.910 (7.1)
<u>6</u> (24R,25S) ^f	0.661	0.921	1.026 (6.26)	0.910 (7.0)	0.886 (6.84)	0.910 (7.0)

Table Methyl Group Region ¹H-NMR (360 MHz) Chemical Shifts^a

^a Given as δ values; J values are given in parentheses in Hz. ^b CDCl₃; Cyclopropyl proton signals: 0.05-0.18 (m, 2H, C-26); 0.45 (m, 1H, C-27); 0.59 (m, 1H, C-25). ^c CDCl₃; 4.94 (ca. d, 2H, C-29); 5.74 (m, 1H, C-26). ^d CDCl₃; 5.171 (m, 2H, C-22/C-23). ^e See ref. 2; CDCl₃; 5.160 (m, 2H, C-22/C-23); 0.835 (d, 3H, C-26). ^f C₆D₆; C-27/C-28 assignments may be reversed.

from <u>3</u> only by a missing C-29 methyl group. In addition the diagnostic peak (m/z 337) from loss of the isopropyl radical in the mass spectrum² of crinosteryl acetate reflects itself in the spectrum of <u>3</u> by loss of a sec-butyl radical (m/z 337). Chemical proof was provided by catalytic hydrogenation of the two isomeric acetates <u>2</u> and <u>3</u> which yielded the same acetate <u>5</u> (¹H-NMR and high resolution mass spectroscopic evidence; GLC rrt 1.71 with cholesteryl acetate = 1.00). As is evident from their ¹H-NMR spectra (see Table), the chemical shifts of the C-27 and C-28 methyl groups in <u>5</u> and aplysteryl acetate <u>6</u> are quite different. This difference can only be due to epimeric substitution at C-25. Since the absolute configuration of aplysterol has been established⁷ by X-ray analysis, the hydrogenation product <u>5</u> should be referred to as 25-epiaplysteryl acetate. This, in turn, leads to the trivial names 26-dehydro-25-epiaplysteryl acetate and 22-dehydro-25-epiaplysteryl acetate for <u>2</u> and <u>3</u> respectively. It could easily be proved by GLCanalysis (rrt 1.59 <u>1</u>; 1.66 <u>2</u>; 1.45 <u>3</u> with cholesteryl acetate = 1.00) that <u>3</u> does not originate from an acid-catalyzed double bond migration in <u>2</u>, or that it was a contaminant in naturally occurring petrosteryl acetate (1).

Based on our earlier mechanistic considerations,² the formation of <u>2</u> may best be explained through the intermediacy of <u>a</u> via a concerted C-C bond fission/hydrogen elimination. The acetate <u>3</u>, which also originates from direct cyclopropane ring-opening and therefore contains the same C-24/C-25 methyl group configuration as the main isomer <u>2</u>, appears to be the product of a 1.5-hydride migration (unprecedented for cyclopropanes) in the initially protonated cyclopropane <u>b</u>₁. On the assumption that intramolecular 1.5-hydride shifts generally proceed via a chairlike transition state,⁸ the intermediate <u>b</u>₁ may be depicted more precisely as <u>b</u>₂ in which the quasi-equatorial methyl group at C-24 additionally stabilizes the six-membered ring conformation. Synthetic experiments with suitably labelled analogs are contemplated in order to provide unambiguous proof for such a hitherto unknown high-order hydride shift among protonated cyclopropanes.



i) HCl (g)/HOAc, rt, I50min; ii) (Ph_3P)_3RhCl/H_2, C_6H_6, rt, I5h; iii) PtO_2/H_2, EtOAc, rt, 3d.







b₂

In summary, the above reported results offer support for the possible operation of an in vivo biomethylation process of the marine sterol side chain which involves isomerization of steroidal cyclopropane intermediates.^{1,5} Furthermore our results represent independent chemical proof that petrosterol (cf. <u>1</u>)^{3a} and aplysterol (cf. <u>6</u>)⁷ possess opposite configurations at C-25.

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